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Specific detection of biomolecules in physiological solutions using graphene transistor biosensors

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Author Contributions: N.G., T.G., and C.M.L. designed research. N.G., T.G. and X.Y. performed experiments. N.G., T.G., X.Y., X.D., W.Z., A.Z. and C.M.L. analyzed and discussed data. N.G., T.G., A. Z., and C.M.L. wrote the paper.

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This article contains supporting information.

Abstract

Nanomaterial-based field-effect transistor (FET) sensors are capable of label-free real-time chemical and biological detection with high sensitivity and spatial resolution, although direct measurements in high ionic strength physiological solutions remain challenging due to the Debye screening effect. Recently, we demonstrated a general strategy to overcome this challenge by incorporating a biomolecule-permeable polymer layer on the surface of silicon nanowire FET sensors. The permeable polymer layer can increase the effective screening length immediately adjacent to the device surface and thereby enable real-time detection of biomolecules in high ionic strength solutions. Here, we describe studies demonstrating both the generality of this concept and application to specific protein detection using graphene FET sensors. Concentration-dependent measurements made with polyethylene glycol (PEG) modified graphene devices exhibited real-time reversible detection of prostate specific antigen (PSA) from 1 to 1000 nM in 100 mM phosphate buffer. In addition, co-modification of graphene devices with PEG and DNA aptamers yielded specific irreversible binding and detection of PSA in pH 7.4 1x phosphate buffered saline (PBS) solutions, whereas control experiments with proteins that do not bind to the aptamer showed smaller reversible signals. In addition, the active aptamer receptor of the modified graphene devices could be regenerated to yield multi-use selective PSA sensing under physiological conditions. The current work presents an important concept toward the application of nanomaterial-based FET sensors for biochemical sensing in physiological environments and thus could lead to powerful new tools for basic research and healthcare.

Significance

Nanoelectronic transistor sensors based on synthesized one- and two-dimensional nanomaterials have achieved real-time label-free detection of a wide-range of biological species with high

sensitivity, although direct analysis of biological samples has been limited due to Debye charge screening in physiological solutions. This paper describes a general strategy overcoming this challenge involving co-modification of the transistor sensor surface with a polymer and receptor, where the polymer forms a permeable layer that increases the effective screening length and receptor enables selective detection of analytes. The capability of this strategy was demonstrated with selective detection of cancer markers in physiological solution, thus opening substantial opportunities for real-time sensing applications in biology and medicine.

Keywords: Field-effect-transistor / Debye screening / surface modification / DNA aptamer receptor / polyethylene glycol (PEG)

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Nanoelectronic biosensors offer broad capabilities for label-free high-sensitivity real-time detection of biological species that are important to both fundamental research and biomedical applications (1-6). In particular, FET biosensors configured from semiconducting nanowires (1, 2), single-walled carbon nanotubes (1, 3, 4) and graphene (1, 5, 6) have been extensively investigated since the first report of real-time protein detection using silicon nanowire devices (7). Subsequent studies have demonstrated highly-sensitive and in some cases multiplexed detection of key analytes, including protein disease markers (8-10), nucleic acids (11-13), and viruses (14), as well as detection of protein-protein interactions (8, 15-17) and enzymatic activity (8).

The success achieved with nanomaterial-based FET biosensors has been limited primarily to measurements in relatively low ionic strength non-physiological solutions due to the Debye-screening length (18, 19). In short, the screening length in physiological solutions, <1 nm, reduces the field produced by charged macromolecules at the FET surface and thus makes real-time label-free detection difficult. The first method reported to overcome this intrinsic limitation of FET biosensors involved desalting to enable subsequent low ionic strength detection (8, 20), although this also precludes true real-time measurements. Truncated antibody receptors (21) and small aptamers (22) also have been used to reduce the distance between target species and the FET surfaces, although the generality of such methods for real-time sensing in physiological conditions requires further study. In addition, recent work has shown that high-frequency mixing-based detection can be used to overcome Debye screening effects (23, 24), although the device geometry may limit this approach in cellular and *in vivo* applications.

Recently, we have developed a strategy to overcome the Debye screening limitation that involves modification of a FET sensor surface with a biomolecule-permeable polymer layer to increase the effective screening length in the region immediately adjacent to the device, and demonstrated this concept for nonspecific detection of PSA using silicon nanowire sensors in physiological solutions (25). To explore the generality of this approach for nanomaterials-based FET sensors and further extend the concept to selective analyte recognition and detection, we herein describe studies demonstrating controlled nonspecific and highly-selective protein detection in physiological media using graphene FET sensors in which the device surfaces are modified only with a biomolecule-permeable polymer layer and co-modified with DNA aptamer/biomolecule-permeable polymer layer, respectively.

Results and Discussion

To realize the biodetection in physiological solutions, a biomolecule-permeable polymer layer was constructed by surface modification as illustrated in Fig. 1A (see *Materials and Methods*). This modification strategy involves (i) adsorption of pyrene butyric acid (PYCOOH) *via* π - π stacking to introduce functional carboxyl groups on the graphene surface (26, 27), followed by (ii) covalent co-coupling of amine-terminated 10 kDa polyethylene glycol (PEG) and either the spacer molecule ethanolamine (ETA) or a DNA aptamer as a specific protein receptor. The coupling procedure (see *Materials and Methods*; (28) uses 1-ethyl-3-[3-dimethylaminopropyl]carbodiimide hydrochloride (EDC) and N-hydroxysulfosuccinimide (Sulfo-NHS) to couple PEG/ETA and PEG/aptamer to the carboxyl groups of PYCOOH-modified graphene device surfaces.

Our sensor chip (Fig. 1B) consists of a FET array with nominally 180 individually addressable graphene devices (see Fig. S1 for full device layout), and was fabricated as follows. First, graphene was synthesized using chemical vapor deposition (CVD) and transferred onto the SiO₂ surface of a Si device fabrication wafer (29). Second, the graphene FET channels were defined by photolithography, and then passivated metal source/drain contacts were fabricated by a second photolithography step, metal thermal evaporation, and sputtered Si₃N₄ (see *Materials and Methods*). Completed device chips were attached and wire-bonded to standard PCB-boards for interfacing to measurement electronics, and a poly(dimethylsiloxane) (PDMS) microfluidic channel was mounted over the central device region for delivery of analyte solutions using a syringe pump (Fig. 1B; see *Materials and Methods*). An optical image (Fig. 1C) shows two types of graphene channels with dimensions of 5×5 μm and 5×10 μm sharing the common source (S) with individually addressable drain (D) contacts.

The properties of functionalized graphene and graphene devices were characterized by several complimentary methods prior to carrying out sensing measurements. First, atomic force microscopy (AFM) studies of PEG-modified graphene devices carried out in 1x PBS (Fig. 2A, see *Materials and Methods*) show a well-defined 6-8 nm step between the PEG-modified graphene FET channel and the SiO₂/Si substrate. Similar AFM measurements on unmodified graphene devices, which show a 0.7-1 nm step, and Raman mapping (see Supplementary text and Fig. S2) are consistent with monolayer graphene (30, 31). These data indicate that the thickness of the PEG layer on the graphene devices is ca. 5-7 nm, which is consistent with previous indentation measurements of PEG layers carried out by AFM (32). In addition, the AFM measurements show that the PEG layer was only observed on graphene, indicating that the

PYCOOH initial modification was specific only to the graphene devices as expected (27), and thus allows for selective functionalization of the graphene sensor surfaces.

Second, device electrical measurements (Fig. 2B-D) highlight several additional points relevant to graphene FET sensors. Conductance versus water-gate voltage data obtained from the same graphene device during sequential modification steps (Fig. 2B; see *Materials and Methods*) shows that the charge neutrality point (CNP) of bare graphene device 0.51 V, increases to 0.60 V and then decreases to 0.25 V after PYCOOH and PEG modification, respectively. CNP measurements recorded from 46 devices during the sequential modification steps (Fig. 2C) yielded average \pm 1SD values of 0.55 ± 0.08 , 0.62 ± 0.07 and 0.22 ± 0.06 for bare, PYCOOH-modified and PEG-modified devices, respectively. These values suggest that the bare graphene was initially p-doped, and that PEG-modification leads to n-doping of the devices (33). This overall doping effect was confirmed by Raman spectroscopy mapping (see Supplementary text and Fig. S2), which showed a 1594 to 1585 cm^{-1} G band shift in bare to PEG-modified graphene. Last, the PEG layer on graphene devices yielded a statistically significant drop ($p < 0.001$, double-sided t-test) of the transconductance, from $228 \pm 65\text{ }\mu\text{S/V}$ before modification to $119 \pm 64\text{ }\mu\text{S/V}$ after completion of the PEG modification (Fig. 2D). These results are consistent with previous observations for PEG-modified silicon nanowire FETs (25), and moreover, an estimate of the effective dielectric constant of the PEG layer (see Supplementary text) supports the hypothesis that this permeable layer reduces the effective dielectric constant at sensor surface compared to aqueous solution.

Initial PSA sensing measurements carried out with ETA and a 1:4 PEG:ETA modified graphene devices in pH 6 phosphate buffer (PB) as a function of solution ionic strength (Fig. 3A) exhibited substantial differences for detection of fixed concentration PSA ($pI = 6.8 - 7.5$; (34,

35). First, measurements made on ETA-modified graphene devices exhibit an easily detected signal at 10 mM PB that rapidly drops to near baseline at 50 mM (black trace, Fig. 3A and Fig. S3). No detectable PSA response was observed in 100 mM PB, which has a ~0.7 nm Debye length (25) comparable with physiological solution. In contrast, PEG/ETA-modified devices show only a gradual decrease in PSA sensing signal with increasing PB concentration. The signal response in 100 mM PB, ca. 14 mV, exceeds the signal recorded from ETA-modified devices in 10 mM PB, and moreover, well-defined sensing signals, ca. 11 mV, can still be recorded in 150 mM PB where the Debye length is ~0.5 nm (25). In addition, these data show that PSA sensing is reversible with the device conductance returning to baseline following addition of pure PB buffer, thus establishing that there is minimal irreversible protein binding to the modified graphene devices.

Protein concentration-dependent sensing experiments carried out on PEG/ETA-modified devices in 100 mM PB (Fig. 3B) demonstrate sensing responses for PSA concentrations from 1 to 1000 nM. A plot of the calibrated sensing signal versus [PSA] recorded simultaneously from three independent devices (Fig. 3C) yields a response varying rapidly at low PSA concentration and then saturating at higher concentrations. Replotting the data as function of log[PSA] (inset, Fig. 3C) defines a relatively linear detection regime $10 \leq [\text{PSA}] \leq 500$ nM. In addition, the concentration dependent sensing data was fit using a Langmuir model (36):

$$S = S_{\max} \times \frac{k \times C}{1 + k \times C} \quad (1)$$

where S and S_{\max} represent the signal and saturation signal, respectively, in response to PSA concentration C , and k is an equilibrium constant. The fit, which has a correlation coefficient of

0.949, yields a value of k , $7.9 \times 10^6 \text{ M}^{-1}$, that is similar to our previous result for concentration-dependent PSA sensing on PEG-modified silicon nanowire sensor (25).

We have also investigated the sensor response for different ionic strength solutions as a function of the PEG:ETA ratio used to modify graphene device surfaces. A summary of results obtained from devices modified with PEG:ETA ratios of 1:2, 1:4, 1:6 and 1:8 (Fig. 3D) demonstrates that graphene devices with 1:4 PEG:ETA modification ratio have the highest sensitivities with signal amplitudes of 21.3 ± 1.1 , 13.7 ± 0.5 and 11.0 ± 0.9 mV in 50, 100 and 150 mM PB, respectively. The sensitivity of devices decreased significantly at higher and lower modification ratios. We hypothesize that the ratio between PEG and the spacer molecule ETA can control the permeability in the PEG layer in terms of dielectric properties and target molecule translocation, although future studies will be needed to determine unambiguously the origin of these results.

Last, we have investigated specific detection of PSA using graphene devices co-modified with PEG and a DNA aptamer for PSA (see *Materials and Methods*). The DNA aptamer is advantageous as the receptor for several reasons, including (i) the conformational changes of highly-charged aptamer upon protein binding (37, 38) can result in a significant change in electric field near the sensor surface even if the solution pH is close to the protein pI, and (ii) the aptamer can be denatured and refolded multiple times without loss of activity (39, 40) for multiple experiments. All experiments were carried out in pH 7.4 1x PBS containing 2 mM Mg^{2+} , where the added Mg^{2+} helps to maintain the active aptamer conformation before and after regeneration (39, 41). Notably, sensing experiments carried out with 1:2 PEG:aptamer co-modified graphene FET devices (blue curve, Fig. 4A) showed a well-defined 5.2 mV irreversible response to 300 nM PSA, which contrasts the response of the PEG/ETA control device (black

curve, Fig. 4A) that exhibited no measurable response. The absence of obvious response in the control device is consistent with the fact that these specific sensing experiments are carried out at physiological pH (vs. pH 6 in Fig. 3), which is close to (vs. lower than) the pI of PSA (34). In control experiments carried out with pure aptamer-modified devices (red curve, Fig. 4A) showed a ca. 1.2 mV irreversible response. This latter sensor response is almost 5-times smaller obtained with the 1:2 PEG:aptamer modified devices, and thus supports the importance of the PEG layer for increasing the effective Debye length and sensitivity of the co-modified graphene FET sensors.

Given the strong aptamer/PSA binding we have investigated regeneration of the active receptor by denaturing the aptamer with guanidinium chloride (42). For example, following initial detection of 100 nM PSA, (black curve, Fig. 4B, 4C) devices were treated with 6 M guanidinium chloride for 10 minutes and then washed in the pH 7.4 buffer. Subsequent detection of the same concentration PSA with the same devices (red curve, Fig. 4B, 4C) showed equivalent sensing signal and confirmed that the functional aptamer was regenerated. In addition, we note that the amplitude of these signals, ca. 2.3 mV, is approximately 2-times smaller than the higher concentration PSA measurement in Fig. 4A, indicating the potential for concentration-dependent detection. Finally, following a second cycle of regeneration, the data in Fig. 4C shows that the device with carcinoembryonic antigen (CEA), which does not bind specifically to the aptamer, yielded a small (near baseline) ~ 0.5 mV reversible binding signal (blue curve, Fig. 4C) in contrast to the irreversible 5-times larger signal obtained for PSA at the same concentration. Together these results show the capability to achieve specific real-time detection of proteins in physiological solution in competition with other proteins.

Conclusions

We have demonstrated a general strategy to enable direct FET sensing measurements in high ionic strength physiological solutions that involves co-modification of device surfaces with PEG and spacer molecules or PEG and aptamer receptors. Concentration-dependent measurements made with PEG/ETA-modified graphene FET devices exhibited real-time reversible detection of PSA from 1 to 1000 nM in 100 mM PB, which has a screening length comparable to physiological solutions, and further showed that detection was possible even at 150 mM PB. In addition, studies carried out using PEG/DNA aptamer modified graphene devices showed irreversible specific binding and detection of PSA in pH 7.4 1x PBS solutions, whereas control experiments with CEA protein, which does not bind specifically to the aptamer, showed smaller reversible signals. In addition, the active aptamer receptor of the modified graphene devices could be regenerated to yield multi-use selective PSA sensing under these physiological conditions. We believe this work represents a critical step toward general application of nanomaterial-based FET sensors in many areas, including *in vitro* and *in vivo* real-time chip-based monitoring of disease marker proteins, which could have substantial impact on both fundamental research and healthcare, as well as integration in free-standing nanoelectronic scaffolds for engineered tissues and *in vivo* implants (43).

Materials and Methods

Graphene synthesis. Monolayer graphene was synthesized on 25 μm thick Cu foil (Alfa Aesar, Tewksbury, MA) *via* a reported low-pressure CVD method (29). Before growth, the Cu foil was electropolished in phosphoric acid (85 wt%) and ethylene glycol (15 wt%) for 30 min and rinsed in DI water. The polished Cu foil was loaded into a 1-inch quartz tube furnace, annealed in 40

sccm H₂ during the 40 min room temperature to 1000 °C heating process, followed by an 20 min anneal at 1000 °C. Graphene growth was initiated by introducing 5 sccm methane into furnace, and growth was continued for 30 min.

FET sensor fabrication. Graphene films were transferred onto Si wafer using a reported poly(methyl methacrylate) (PMMA) method (29), where PMMA (PMMA-C5, Microchem Corp., Newton, MA) was spin-coated on as-grown graphene/Cu foil, at 2000 rpm for 1 min, the Cu foil was etched in ammonium persulfate aqueous solution (10 wt%), the floating PMMA/graphene film was rinsed in DI water, and then transferred to the SiO₂ surface of Si/SiO₂ target wafer. PMMA was dissolved in acetone at 70 °C.

The graphene was patterned using photolithography. (i) LOR 3A (Microchem Corp., Newton, MA) was spin-coated onto graphene/Si wafer, at 4000 rpm for 1 min, followed by 180 °C baking for 2 min. S1805 (Microchem Corp., Newton, MA) was spin-coated on the wafer at 4000 rpm for 1 min, followed by 1 min of 115 °C baking. (ii) Graphene channels were defined by photolithography. (iii) After developing in CD-26 (Microchem Corp., Newton, MA) for 1 min, the graphene/Si wafer was placed into an O₂ plasma cleaner (Fetmo, Diener electronic GmbH + Co., Germany) to etch graphene in regions without photoresist protection. (iv) Photoresist on the graphene/Si wafer was removed in Remover PG (Microchem Corp., Newton, MA). (v) The patterned graphene was further cleaned by vacuum thermal annealing at 300 °C for 1 hour (Jipelec rapid thermal processor, SEMCO Technologies, France) to reduce PMMA residue (44).

Device fabrication was completed by combination of photolithography with metal and Si₃N₄ deposition. (i) LOR 3A (Microchem Corp., Newton, MA) was spin-coated onto graphene/Si wafer, at 4000 rpm for 1 min, followed by 180 °C baking for 2 min. S1805 (Microchem Corp.,

Newton, MA) was subsequently coated on top, at 4000 rpm for 1 min, followed by 1 min 115 °C baking. (ii) Metal contacts were defined using photolithography. (iii) After developing in CD-26 for 1 min, 10 nm Cr/65 nm Au/3 nm Cr were deposited *via* thermal evaporation, with deposition rates of 0.2 Å/s, 2.0 Å/s and 1 Å/s, respectively. (iv) A 30 nm Si₃N₄ passivation layer was deposited over the metal contacts *via* magnetron sputtering (Orion 3, AJA international Inc., Scituate, MA).

Electrical characterization. Devices chips were wire-bonded to a PCB interface board as shown in Fig. 1B, and then a PDMS microfluidic channel was mounted on the sensor chip with the channel aligned with the central region of the device chip where the graphene FETs are located. Polyethylene tubing was attached to the inlet and the outlet holes on the PDMS microfluidic channel, and buffer or protein/buffer solutions were drawn through the channel using a syringe pump.

Water-gate versus conductance measurements were carried out in 1x PBS using a home-made probe station. The water-gate was varied at 100 mV/s sweep rate while monitoring the graphene FET conductance for a fixed 100 mV source/drain voltage; the FET conductance was amplified (1211; DL Instruments, LLC, Ithaca, NY). The resulting conductance versus water-gate curves are used to calculate CNP and transconductance values for the devices.

Depending on the integrity of transferred graphene film, the yield of working sensors (showing water-gate response) calculated based on all of 180 available source/drain electrodes on each sensor chip was usually above 95%.

AFM characterization. The device chip was mounted in an airtight chamber, before being immersed in a droplet of 1x PBS, and measurements were carried out at room temperature using

an AFM (MFP-3D Coax AFM, Oxford Instruments Asylum Research, Inc., Santa Barbara, CA). A Si AFM tip (AC160TS-R3-35, Oxford Instruments Asylum Research, Inc., Santa Barbara, CA) was used to acquire both AFM height and amplitude images of graphene channel, under constant amplitude tapping (AC) mode.

Surface modification. The device chip was soaked in 0.6 mM 1-pyrenebutyric acid (PYCOOH, Sigma-Aldrich, St Louis, MO) in dimethylformamide (DMF, Sigma-Aldrich, St Louis, MO) solution for an hour. After washing with DMF 3 times, the device chip was immersed in pure DMF for 1 hour at 60 °C with agitation to further remove excess PYCOOH. Then, the device chip was modified with ethanolamine (ETA, 411000, Sigma-Aldrich, St Louis, MO), pure aptamer, the mixture of 10 kDa PEG (PSB-267, Creative PEGWorks, NC) and ETA, or PEG and aptamer (5'-NH₂-C₆-TTTTTAATTAAAGCTCGCCATCAAATAGCTTT-3', Gene Link, NY) using 1-ethyl-3-[3-dimethylaminopropyl]carbodiimide hydrochloride (EDC, 03449, Sigma-Aldrich, St Louis, MO) and N-hydroxysulfosuccinimide sodium salt (56485, Sigma-Aldrich, St Louis, MO) cross-linkers (28). For the ETA and PEG/ETA modified graphene devices, the functionalization was carried out in a petri dish on a shaker for 2 hours, followed by rinsing with DI water. For pure aptamer and PEG/aptamer and the corresponding ETA/PEG (control) modified graphene devices, the functionalization was conducted in the PDMS microfluidic channel for 1.5 hours, followed by rinsing with pH 7.4 1x PBS buffer containing 2 mM Mg²⁺.

Aptamer regeneration. To release PSA and regenerate the aptamer receptors, the device chip was incubated with 6 M guanidinium chloride (G3272, Sigma-Aldrich, St Louis, MO) for 10 minutes, followed by DI water wash and then PBS (containing 2 mM Mg²⁺) buffer to reconstitute the functional conformation of PSA aptamers.

Sensing measurements. Measurements were carried out using up to 3 independent lock-in amplifiers (SR830, Stanford Research Systems, Inc. Sunnyvale, CA) with 30 mV modulation amplitudes and modulation frequencies of 79, 97, and 103 Hz to simultaneously record 3 graphene devices selected from 180 devices on the chip. A Ag/AgCl electrode was used as a reference; sensing experiments were all carried out in the linear regime, which in some cases required a DC offset of the water-gate/reference voltage. The conductance versus time data were digitized and recorded on computer using custom software. The water-gate responses of devices were characterized before PSA detection experiments, and device transconductance values were determined from the water-gate data. Graphene FET signals were converted to absolute millivolt (mV) values for the calibrated sensing signals using the device transconductance determined from these water-gate measurements.

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Figure Legends

Figure 1. Active sensor surface and sensor chip. (A) Illustration of a graphene FET device with co-modification of PEG and a small molecule spacer or PEG and a receptor, for nonspecific and specific detection of the analyte, respectively. Dark purple: Silicon wafer, light blue: 600 nm SiO₂, black: graphene, yellow: metal contact, light purple: Si₃N₄ passivation layer, red oval: PY-COOH, green: spacer molecule, blue: PEG chain, red star: analyte. EDC and Sulfo-NHS serve as crosslinkers to couple PYCOOH with PEG and spacer molecules. (B) Optical image of a typical device chip (central green square) mounted on a PCB interface board that is plugged into the input/output interface connected to a computer controlled data acquisition system. The copper squares surrounding the device chip are connected to the chip by wire-bonding. A PDMS microfluidic channel is mounted onto the central graphene region. The inlet/outlet of solution was controlled with a syringe pump *via* tubing. Scale bar: 1 cm. (C) Bright-field microscopy image of two types of graphene channels with dimensions of 5×5 μm and 5×10 μm, sharing the common source (S) with individually addressable drain (D) contacts. The white dashed rectangle highlights one graphene FET. Scale bar: 20 μm.

Figure 2. Characterizations of modified graphene surfaces. (A) AFM image shows a ~ 6-8 nm PEG layer in 1x PBS on the graphene FET channel with respect to the SiO₂/Si substrate. Scale bar: 1 μm. (B) Conductance versus water-gate voltage data recorded from a typical device (source/drain voltage = 100 mV) in 1x PBS before and after sequential modification steps: bare graphene (red), and PY-COOH- (green) and PEG-modified graphene (blue). (C) Box plots of CNPs on 46 bare graphene (red), 46 PY-COOH-modified (green) and 37 PEG-modified graphene devices (blue). (D) Box plots of the transconductance before and after PY-COOH

(green) and PEG (blue) modifications. The highest and lowest horizontal lines in the boxes in (C) and (D) represent the standard deviation, while the middle line represents the mean value. The vertical whiskers show the maximum and minimum values.

Figure 3. Nonspecific PSA detection in high ionic strength solutions. (A) PB concentration-dependent PSA signal amplitude versus time data recorded from ETA-modified (control) and ETA/PEG-modified devices. The black trace represents the response of the ETA-modified devices. The PSA concentration in all experiments shown in (A) was 100 nM. (B) Time-dependent signal response traces at different PSA concentrations for a PEG-modified graphene FET sensor in 100 mM PB. (C) Plot of the sensor response versus PSA concentration. The red line is fit of the data with Langmuir adsorption isotherm with $k = 7.9 \times 10^6 \text{ M}^{-1}$. The inset shows the sensor response (mV) versus logarithm of the PSA concentration. The PEG: ETA modification ration in A, B and C was 1:4. (D) Signal amplitude dependence on the ratio between PEG and ETA in the modification layer. All experiments were carried out below the PSA isoelectric point in pH 6 PB. The error bars in (C) and (D) represent the standard deviation from three independent devices on the same device chip.

Figure 4. Specific PSA detection in high ionic strength solutions. (A) 300 nM PSA responses from 1:2 PEG:ETA-, pure aptamer- and 1:2 PEG :aptamer-modified graphene FET devices. (B) A device with 1:2 PEG:aptamer modification shows consistent response to 100 nM PSA before and after 6 M guanidinium chloride regeneration for 10 minutes. (C) A device (distinct from B) with 1:2 PEG: aptamer modification first shows consistent response to 100 nM PSA before and

after regeneration, and then shows a weaker and reversible response to 100 nM CEA. All experiments were carried out in pH 7.4 1x PBS containing 2mM Mg^{2+} .







